

EFFECT OF P-AMINOSALICYLIC ACID (PAS) ON THE LOSS OF ACID-FASTNESS PRODUCED IN TUBERCLE BACILLI BY ISONIAZID.

K. G. VARMA.

(*From the Tuberculosis Chemotherapy Centre, Madras.**)

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The microbiological assay method for estimating free isoniazid in serum described by Mandel *et al.* (1956), used as the assay end-point the dilution producing loss of acid-fastness in 50 per cent of bacilli. The advantage of this end point is that loss of acid-fastness is produced specifically by isoniazid, and not by either PAS (Mandel *et al.*, *loc. cit.*) or streptomycin (Middlebrook, 1952). Mandel *et al.* (*loc. cit.*) claimed that the addition of as much as 1,000 $\mu\text{g./c.c.}$ of PAS to undiluted human sera did not interfere with the assay of isoniazid in the serum using loss of acid-fastness as the end-point. Since Mandel *et al.* (*loc. cit.*) did not however, describe their results fully, it remained a possibility that PAS in concentrations near that necessary to inhibit the growth of tubercle bacilli could act synergistically or antagonistically with isoniazid in producing loss of acid-fastness. This was investigated in the following experiments.

METHOD.

The method used for investigating the effect of PAS on the loss of acid-fastness produced by isoniazid was similar to that used for the microbiological assay of isoniazid (Mandel *et al.*, *loc. cit.*). A series of twofold increasing concentrations of PAS from 0.25 to 2 $\mu\text{g./c.c.}$ of sodium PAS without isoniazid or with 0.02, 0.03, 0.04 or 0.5 $\mu\text{g./c.c.}$ isoniazid were prepared in 2 c.c. volumes of 7H-10 liquid medium (Cohn *et al.*, 1954 ; Gangadharam, Selkon and Bhatia, 1961), but without Tween 80 or glycerol. Each tube was inoculated with 0.1 c.c. of an 8-day old culture of H37Rv grown in 7H-10 medium containing 0.05 per cent Tween 80. After incubation at 37° C. for five days, smears were prepared from the deposits of growth and stained by the Ziehl-Neelsen method.

RESULTS.

The percentage of acid-fast bacilli was first determined by examining six oil-immersion fields and estimating approximately the percentage of acid-fast bacilli among those seen (Table I). The loss of acid-fastness in the series of PAS concentrations with the isoniazid concentration which produced approximately 50 per cent loss of acid-fastness, namely 0.03 $\mu\text{g./c.c.}$, was then more accurately determined by actually counting the acid-fast, non-acid-fast and doubtfully acid-fast bacilli. Each estimation was based on a count of at least 100 bacilli. The experiment was carried out in duplicate and the average of the results are given in Table II.

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TABLE I.

Percentage of acid-fats bacilli in different concentrations of isoniazid and PAS.

Concentration of INAH in $\mu\text{g./c.c}$	CONCENTRATIONS OF SODIUM PAS DIHYDRATE IN $\mu\text{g./c.c}$.				
	0.00	0.25	0.5	1.0	2.0
0.00	100	100	100	100	100
0.02	80	80	70	70	70
0.03	40	40	35	35	30
0.04	*	5	5	5	5
0.05	0	0	0	0	0

* Contaminated.

TABLE II

The effect of different PAS concentrations on the loss of acid-fastness produced by 0.03 $\mu\text{g./c.c}$. of isoniazid.

Staining character.	CONCENTRATIONS OF SODIUM PAS DIHYDRATE IN $\mu\text{g./c.c}$.				
	0.00	0.25	0.5	1.0	2.0
Acid-fast	39.8	27.2	34.8	26.8	41.6
Non-acid-fast	51.7	70.2	57.5	69.3	55.8
Doubtful	8.5	2.6	7.7	3.9	3.6

* Percentage of acid-fast, non-acid fast and doubtfully acid-fast bacilli as determined by actual count.

From the above results, it can be conclude that PAS in twofold increasing concentration from 0.25 $\mu\text{g./c.c}$ to 2.0 $\mu\text{g./c.c}$ of the sodium salt had no effect on the loss of acid-fastness produced by isoniazid.

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